

Kinetic chemistry (321 C) for 3rd Year Chemistry Students

- [1] When the volume of the lungs increase, the air pressure decreases. It is an application of.....
- [2] In Boltzmann distribution of molecular speeds, the speed scale is
- [3] The relation between the average (av), the most probable (mp) and the root mean square (rms) speeds is
- [4] The average kinetic energy per mole equals
- [5] The pressure of an ideal gas equals to
- [6] Avogadro's Law illustrate the relationship between